chapter02

Section Outline

Section 2-1

2-1 The Nature of Matter

- A. Atoms
- B. Elements and Isotopes
- 1. Isotopes
- 2. Radioactive Isotopes
- C. Chemical Compounds
- D. Chemical Bonds
- 1. Ionic Bonds
- 2. Covalent Bonds
- 3. Van der Waals Forces

An Element in the Periodic Table

Section 2-1

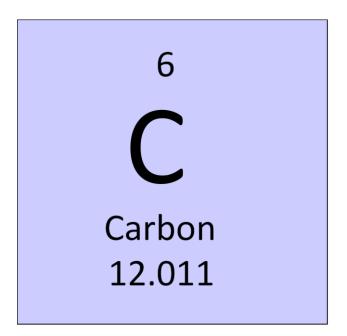


Figure 2-2 Isotopes of Carbon

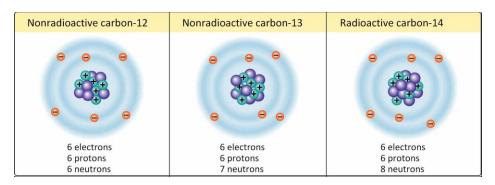


Figure 2-3 Ionic Bonding

Section 2-1

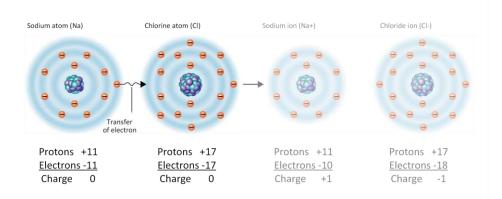


Figure 2-3 Ionic Bonding

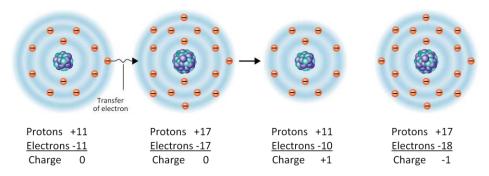
Section 2-1

Sodium atom (Na)

Chlorine atom (Cl)

Sodium ion (Na+)

Chloride ion (Cl-)



Section Outline

Section 2-2

- 2–2 Properties of Water
- A. The Water Molecule
- 1. Polarity
- 2. Hydrogen Bonds
- B. Solutions and Suspensions
- 1. Solutions
- 2. Suspensions
- C. Acids, Bases, and pH
- 1. The pH Scale
- 2. Acids
- 3. Bases
- 4. Buffers

pH Scale

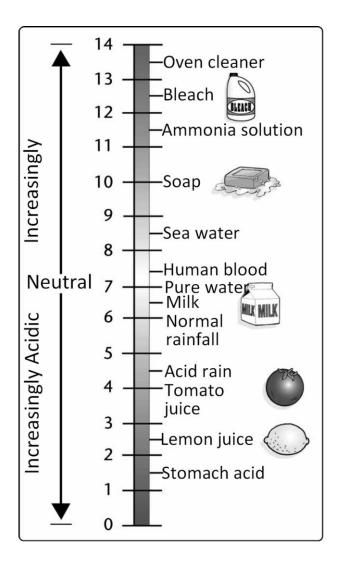


Figure 2-9 NaCl Solution

Cl-

Water

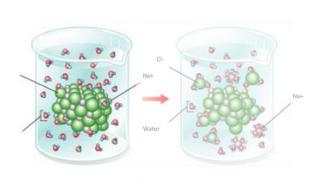
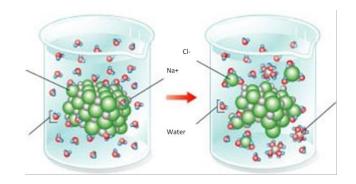


Figure 2-9 NaCl Solution

Cl-

Water



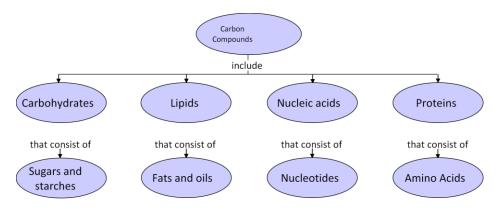
Na+

Section Outline

Section 2-3

- 2–3 Carbon Compounds
- A. The Chemistry of Carbon
- B. Macromolecules
- C. Carbohydrates
- D. Lipids
- E. Nucleic Acids
- F. Proteins

Concept Map



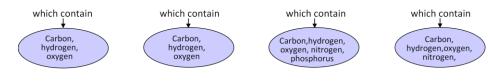


Figure 2-11 Carbon Compounds

Section 2-3

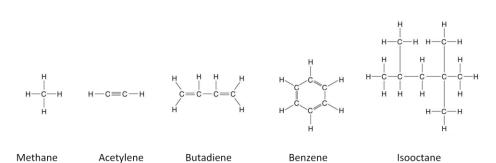


Figure 2-13 A Starch

Section 2-3

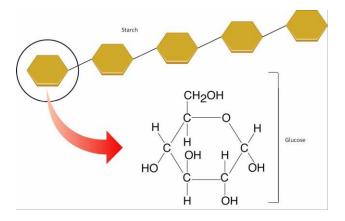


Figure 2-16 Amino Acids

Section 2-3

Amino group

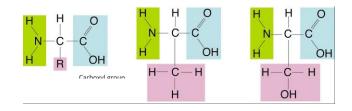


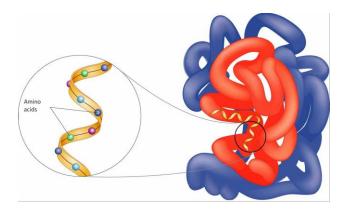
Figure 2-16 Amino Acids

Section 2-3

Amino group

Figure 2-17 A Protein

Section 2-3



1 - Nucleic Acid

Nucleic Acid

Section Outline

Section 2-4

- 2–4 Chemical Reactions and Enzymes
- A. Chemical Reactions
- B. Energy in Reactions
- 1. Energy Changes
- 2. Activation Energy
- C. Enzymes
- D. Enzyme Action
- 1. The Enzyme-Substrate Complex
- 2. Regulation of Enzyme Activity

Effect of Enzymes

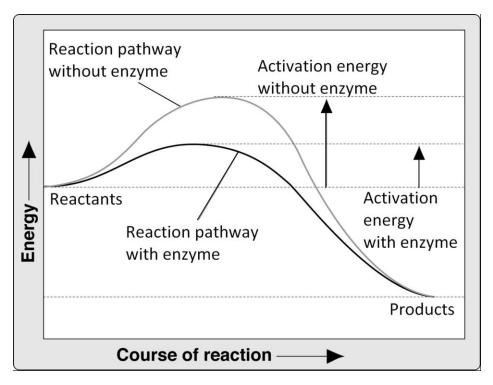


Figure 2-19 Chemical Reactions

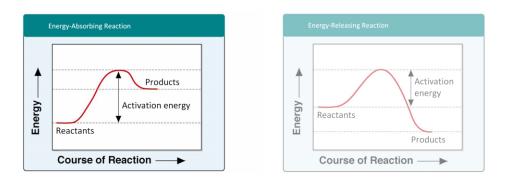


Figure 2-19 Chemical Reactions

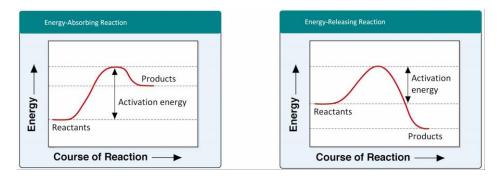
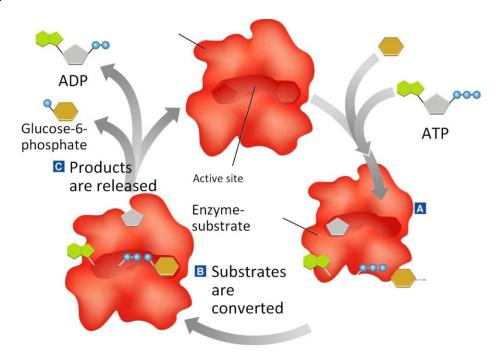


Figure 2-21 Enzyme Action

Section 2-4



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